35.1.18

AOAC Official Method 937.09 Salt (Chlorine as Sodium Chloride) in Seafood

Volumetric Method First Action 1937 Final Action

A. Reagents

- (a) Silver nitrate standard solution.—0.1M. Prepare as in **941.18A** (see A.1.11) and standardize against 0.1M NaCl containing 5.844 g of pure dry NaCl/L.
- (b) Ammonium thiocyanate standard solution.—0.1M. Prepare as in **941.18D(b)** (see A.1.11) and standardize against 0.1M AgNO₃.
 - (c) Ferric indicator.—Saturated solution of FeNH₄(SO₄)₂·12H₂O.

B. Determination

- (a) Shellfish meats.—Weigh 10 g meats, liquid, or mixed meats and liquid, into 250 mL Erlenmeyer or beaker.
- (b) Other fish products.—Use suitable size test sample, depending on NaCl content.

Add known volume 0.1M AgNO $_3$ solution, more than enough to precipitate all Cl as AgCl, and then add 20 mL HNO $_3$. Boil gently on hot plate or sand bath until all solids except AgCl dissolve (usually 15 min). Cool, add 50 mL H $_2$ O and 5 mL indicator, and titrate with 0.1N NH $_4$ SCN solution until solution becomes permanent light brown. Subtract mL 0.1M H $_4$ SCN used from mL 0.1M AgNO $_3$ added and calculate difference as NaCl. With 10 g test sample each mL 0.1N AgNO $_3$ = 0.058% NaCl.

References: JAOAC 20, 410(1937); 23, 589(1940).

CAS-7647-14-5 (sodium chloride)